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Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.002 Å R factor = 0.033 wR factor = 0.095 Data-to-parameter ratio = 17.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

A monoclinic polymorph of isoxsuprine hydrochloride

A monoclinic polymorph of the title compound [systematic name: 2-hydroxy-2-(4-methoxyphenyl)-1-methyl-N-(1-phenoxy-2-propyl)ethanaminium chloride], $C_{18}H_{24}N_2O_3^+$ ·Cl⁻, has been found in addition to the already known triclinic polymorph. The molecular conformation in both polymorphs is very similar.

Comment

The title compound, (I), is used as a vasodilator and in the treatment of navicular disease. A triclinic polymorph, (II), has already been described (Léger *et al.*, 1981). It crystallizes in $P\overline{1}$ with two molecules in the asymmetric unit. We present here a monoclinic polymorph in $P2_1/c$ with Z = 4. A perspective view of (I) is shown in Fig. 1.



The cells of the two polymorphs have nothing in common, except the volume. The unit-cell axes and interaxial angles are completely different. The molecular conformations, on the other hand, are essentially the same. A least-squares fit of (I) with (II) gives r.m.s. deviations of 0.228 and 0.200 Å for the two molecules in the asymmetric unit of (II) (Fig. 2).

The crystal packing is stabilized by several $N-H\cdots Cl$ and $O-H\cdots Cl$ hydrogen bonds (Table 1).

Experimental

The sample of the title compound was obtained as a gift from Jayco Chemical Industries, India. It was used without further purification and recrystallized from methanol.

Crystal data	
$C_{18}H_{24}NO_{3}^{+}\cdot Cl^{-}$	$D_x = 1.290 \text{ Mg m}^{-3}$
$M_r = 337.83$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 24 065
a = 12.0767 (11) Å	reflections
b = 19.3499 (12) Å	$\theta = 3.7-27.6^{\circ}$
c = 7.8018 (7) Å	$\mu = 0.23 \text{ mm}^{-1}$
$\beta = 107.482 \ (7)^{\circ}$	T = 173 (2) K
$V = 1738.9 (3) \text{ Å}^3$	Block, colourless
Z = 4	$0.44 \times 0.39 \times 0.35 \text{ mm}$

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Data collection

Stoe IPDS-II two-circle diffractometer ω scans Absorption correction: multi-scan (MULABS; Spek, 2003; Blessing, 1995) $T_{\min} = 0.904, T_{\max} = 0.923$ 21 164 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.033$ $wR(F^2) = 0.095$ S = 1.074008 reflections 224 parameters H atoms treated by a mixture of independent and constrained refinement

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$	
O1-H1···Cl1 ⁱ	0.78 (2)	2.36 (2)	3.1285 (11)	168.3 (18)	
N3−H3A···Cl1 ⁱⁱ	0.895 (16)	2.529 (16)	3.3726 (10)	157.2 (13)	
$N3-H3B\cdots Cl1$	0.844 (17)	2.344 (17)	3.1852 (11)	174.3 (14)	
$O14-H14\cdots Cl1^{iii}$	0.84 (3)	2.33 (3)	3.1730 (11)	178 (2)	

Symmetry codes: (i) x, y, z - 1; (ii) $x, \frac{1}{2} - y, z - \frac{1}{2}$; (iii) 1 - x, 1 - y, 1 - z.

H atoms bonded to carbon were refined with fixed individual displacement parameters $[U_{iso}(H) = 1.2U_{eq}(C) \text{ or } 1.5U_{eq}(C_{methyl})]$ using a riding model, with C-H = 1.00, 0.99, 0.98 and 0.95 Å for tertiary CH, secondary CH, methyl and aromatic CH, respectively. H atoms bonded to nitrogen and oxygen were refined freely.

Data collection: X-AREA (Stoe & Cie, 2001); cell refinement: X-AREA; data reduction: X-AREA; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: XP in SHELXTL-Plus (Sheldrick, 1991); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2003).

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 $w = 1/[\sigma^2(F_o^2) + (0.0585P)^2]$ + 0.1915P] where $P = (F_0^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\rm max} = 0.001$ $\Delta \rho_{\rm max} = 0.27 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\rm min} = -0.26 \text{ e } \text{\AA}^{-3}$

Figure 1

Perspective view of the title compound with the atom numbering; displacement ellipsoids are drawn at the 50% probability level. The dashed line indicates a hydrogen bond.

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Figure 2

Least-squares fit of (I) (dashed line) with one of the two molecules (solid line) in the asymmetric unit of (II).

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